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SEPARATION OF VITAMIN D₃ METABOLITES AND THEIR ANALOGUES BY HIGH-PRESSURE LIQUID CHROMATOGRAPHY*

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SUMMARY

The separation of vitamin D_3 metabolites and their analogues by high-pressure liquid chromatography was investigated using a Zorbax SIL column. Each metabolite can be separated effectively with a solvent consisting of 2% methanol-methylene chloride or gradient elution from 0.02% to 6% methanol-methylene chloride. The C-24 epimers of 1 α ,24-dihydroxy- and 1 α ,24,25-trihydroxy-vitamin D_3 are resolved as their free forms, while the epimers of 24-hydroxy- and 24,25-dihydroxy-vitamin D_3 can be separated as their trimethylsilyl derivatives.

INTRODUCTION

During the past decade, it has been well established, mainly by DeLuca and co-workers, that vitamin D_3 (D_3) must first be metabolized in the liver to 25-hydroxyvitamin $D_3[25-(OH)-D_3]$ and subsequently in the kidney to $1\alpha, 25$ -dinydroxy-vitamin $D_3 \left[1\alpha, 25 - (OH)_2 - D_3 \right]$ in order to carry out its well known functions involving the stimulation of intestinal calcium transport, bone calcium mobilization and calcification of bone. Other D_3 metabolites, 24,25- and 25,26-dihydroxy-vitamin D_3 [24,25- and 25,26-(OH)₂-D₃ and $1\alpha,24,25$ -trihydroxy-vitamin D₃ $[1\alpha,24,25$ -(OH)₃-D₃], have also been isolated and their structures and biological activities determined by DeLuca¹. However, the biological significance and the stereochemistry of the 24hydroxyl group of these compounds have not been clarified. During the course of our studies of the synthesis of the D_3 metabolites and on the determination of the configuration of the 24-hydroxyl group, it became necessary to develop a separation method that has high resolution and sensitive detection system. A study was therefore carried out of the separation of D_3 metabolites and their analogues by high-pressure liquid chromatography (HPLC). The compounds studied are shown in the structure below.

* This paper is Part 34 in the series of "Studies on Steroids". For Part 33, see L. Chardon-Loriaux, M. Morisaki and N. Ikekawa, *Phytochemistry*, in press.

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	R ₂ R ₃	R	R2	R3	R.		
\rightarrow	$\overset{R_2}{\swarrow}\overset{R_3}{\overset{R}}{\overset{R_3}{\overset{R_3}{\overset{R}}{\overset{R_3}{\overset{R}}{\overset{R}}{\overset{R}}{\overset{R}}{\overset{R}}{\overset{R}}{\overset{R}}{\overset{R}}{\overset{R}}{\overset{R}}{\overset{R}}{\overset{R}}{\overset{R}}{\overset{R}}{\overset{R}}{\overset{R}}{\overset{R}}}{\overset{R}}}{\overset{R}}}{\overset{R}}}{\overset{R}}}}}}}}$	H	H	14 IV T 14 MT	H		
	"4 1α-(Ο]	H)-D, OH	H	H	H	e de la constante de la consta La constante de la constante de	
	24 <i>R-</i> ((DH)-D ₃ H	OH	H	H		
	245-(0)H)-D ₃ H	Н	OH	H		
	25-(OI	H)-D ₃ H	H	H	OH		
	1a,241	R-(OH)2-D3 OH	OH	H	H		
	1a,245	G-(OH)2-D3 OH	H	OH	H		
a de la companya de	1α,25-	(OH)2-D3 OH	H	H	OH		
	24 <i>R</i> ,25	5-(OH) ₂ -D ₃ H	OH	H	OH		
	245,25	-(OH) ₂ -D ₃ H	H	OH	OH		
HO R1	1α,241	R,25-(OH)2-D3 OH	OH	H	OH		
	1α,245	5,25-(OH)3-D3 OH	H	OH	OH		

TABLE I

RETENTION TIMES OF VITAMIN D $_3$ METABOLITES AND THEIR ANALOGUES IN HIGH PRESSURE LIQUID CHROMATOGRAPHY ON ZORBAX SIL

Compound	Retention time (min)	Solvent	Pressure* (kg/cm ²)
D ₃	3.7	Gradient elution:	93
24 <i>R</i> -(OH)-D ₃	10.1	0.02% MeOH-	•
24S-(OH)-D3	10.1	CH ₂ Cl ₂ , gradient rate	
25-(OH)-D ₃	10.7	0.3%/min, to 6% MeOH-	
1a-(OH)-D ₃	12.2	CH ₂ Cl ₂	÷
24R,25-(OH)2-D3	13.4		
24S,25-(OH)2-D3	13.4		·
$1\alpha, 24R-(OH)_2-D_3$	15.4		
1a,24S-(OH)2-D3	15.6		
1a,25-(OH)-D3	16.9		
1a,24R,25-(OH)3-D2	21.0		-
1a,24S,25-(OH)3-D3	21.2		· · · · ·
D_3	2.3	2% MeOH-CH2Cl2	90
24R-(OH)-D3	3.0		•
24S-(OH)-D3	3.0		
25-(OH)-D ₃	3.8		
1a-(OH)-D3	6.2		
24R,25-(OH)2-D3	7.7	1. State 1.	
245,25-(OH)2-D3	7.7		
1a,24R-(OH)2-D3	13.1		
1a,24S-(OH)2-D3	13.9		
1a,25-(OH)2-D3	19.2		
24 <i>P</i> -(OH)-D ₃ di-TMS	13.8	2% CH2Cl2-n-hexane	93
245-(OH)-D3 di-TMS	12.1 -		
24R,25-(OH)2-D3 tri-TMS	21.2	2% CH,Cl,-n-hexane	83
245,25-(OH)2-D3 tri-TMS	18.4	· · · · · · · · · · · · · · · · · · ·	
1z,24R-(OH)2-D3	16.9	1.5% MeOH-n-hexane	92
1α,24S-(OH) ₂ -D ₃	18.0		
1a,24R,25-(OH)1-D1	12.5	3.5% MeOH-CH2Cl2	90
1a,245,25-(OH) ₃ -D ₃	13.2	STA NO MOULE CLIZON	

* Flow-rate 0.4-0.42 ml/min.

HPLC OF VITAMIN D3 METABOLITES

EXPERIMENTAL

D₃ derivatives

All of the samples of D₃ metabolites and their analogues were synthetic compounds: 25-(OH)- (ref. 2), 1α -(OH)- (refs. 3 and 4), 24R-(OH)- (ref. 5), 24S-(OH)-(ref. 5), 1α ,25-(OH)₂- (refs. 6 and 7), 1α ,24*R*-(OH)₂- (ref. 8), 1α ,24*S*-(OH)₂- (ref. 8), 24R,25-(OH)₂- (refs. 9 and 10), 24S,25-(OH)₂- (refs. 9 and 10), 1α ,24*R*,25-(OH)₃-(ref. 11), and 1α ,24*S*,25-(OH)₃-D₃ (ref. 11).

Trimethylsilyl (TMS) derivatives

In a small PTFE-capped tube, 10 μ g of hydroxy-D₃ were dissolved in 50 μ l of dry *n*-hexane plus 10 μ l of trimethylsilylimidazole. After heating at 70° for 10 min, or at 50° for 20 min, 0.3 ml of water was added, the mixture was extracted with *n*-hexane and the extract was injected into the chromatograph. The TMS group was removed by treating the TMS ether (10 μ g) with 0.1 ml of 1% potassium hydroxide in methanol at room temperature for 2 h. The free D₃ was extracted with ethyl acetate.

Instrument

A Shimadzu-DuPont Model 830 liquid chromatograph equipped with a UV detector (254 nm) and a gradient system was used, with a Zorbax SIL column, 25 cm \times 2.1 mm I.D. The separation conditions are given in Table I.

RESULTS AND DISCUSSION

Separation of D_3 metabolites and their analogues

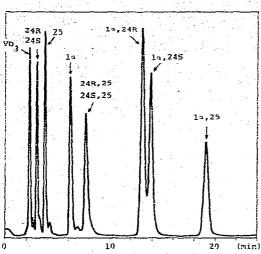
In comparison with silicic acid or Sephadex column chromatography, which have been employed for the separation of the metabolites¹², a higher resolution was obtained by HPLC. After testing several types of column, we found that Zorbax SIL exhibited the best resolution for this purpose.

 D_3 , 25-(OH)- D_3 , 24,25-(OH)₂- D_3 and 1α ,25-(OH)₂- D_3 can easily be separated from each other with a solvent consisting of 2% methanol-methylene chloride (Fig. 1). By gradient elution with a mobile phase of 0.02-6% methanol in methylene chloride, all of the metabolites so far isolated and their analogues, 1α -(OH)-, 24-(OH)and 1α ,24-(OH)₂- D_3 , can be separated, as shown in Fig. 2.

These results illustrate the usefulness of HPLC for the metabolic studies of D_3 and its hydroxy analogues, which has also been reported by Jones and DeLuca¹³.

Separation of the epimers of the 24-hydroxyl group

In previous papers^{9,10}, the synthesis of 24*R*- and 24*S*-24,25-(OH)₂-D₃ was reported. For the identification of the metabolite 24,25-(OH)₂-D₃ isolated by Holick *et al.*¹⁴, it was necessary to find a method for separating the epimers. As shown in Figs. 1 and 2 and Table I, the 24*R* and 24*S* isomers of 24-(OH)- and 24,25-(OH)₂-D₃ exhibited identical retention times but of 1α ,24-(OH)₂- and 1α ,24,25-(OH)₃-D₃ had slightly different retention times. It was found that the TMS derivatives of the 24-epimers of 24,25-(OH)₂-D₃ could be separated using the solvent methylene chloride-*n*-hexane (Fig. 3). The TMS derivative was also resolved by TLC using silica gel. This



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Fig. 1. Separation of vitamin D_3 (VD₃) metabolites and their analogues on Zorbax SIL with 2% methanol-methylene chloride.

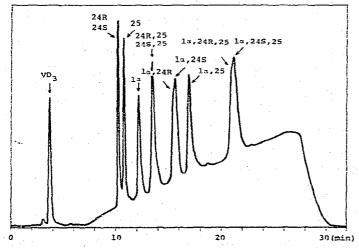


Fig. 2. Separation of vitamin D_3 (VD₃) metabolites and their analogues on Zorbax SIL with gradient elution from 0.02 to 6% methanol in methylene chloride.

derivative was selected because of its easy derivatization and removal. By this method, the configuration of the 24-hydroxyl group of natural $24,25-(OH)_2-D_3$ was determined to be R, using co-chromatography with the ³H-labelled compound obtained biologically¹⁵.

The epimers of 24-(OH)-D₃ can be also separated as their TMS derivatives (Table I). This method was applied to the separation of R and S epimers of [24-³H]-24-(OH)-D₃ derived from 24-oxo-D₃ by sodium [³H] borohydride reduction¹⁶.

On the contrary, TMS derivatives of 24-epimers of 1α ,24-(OH)₂-D₃ and 1α ,24, 25-(OH)₃-D₃ had the same retention time, but those epimers could be separated as their free forms as shown in Fig. 4 and Table I. The 24R isomer was eluted faster than the 24S isomer, while with the TMS derivatives of 24-(OH)- and 24,25-(OH)₂-D₃

HPLC OF VITAMIN D, METABOLITES

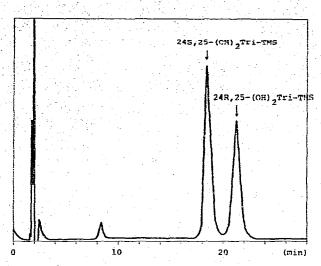


Fig. 3. Separation of $24R_{25}-(OH)_2-D_3$ tri-TMS and $24S_{25}-(OH)_2-D_3$ tri-TMS on Zorbax SIL with 2% methylene chloride-*n*-hexane.

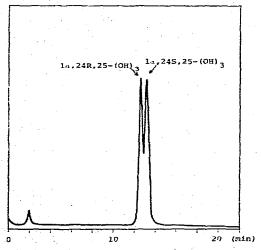


Fig. 4. Separation of 1α ,24*R*,25-(OH)₃-D₃ and 1α ,24*S*,25-(OH)₃-D₃ on Zorbax SIL with 3.5% methanol-methylene chloride.

the reverse elution order was observed. It is interesting to note the effect of the 1a-hydroxyl group on the separation of the stereoisomers at the side-chain of steroids.

For the synthesis of vitamin D_3 analogues, usually the hydroxycholesterols are converted into the D_3 form via a 5,7-diene system. In the synthetic route to the 24R and 24S isomers of 24-hydroxy- D_3 analogues, the C-24 epimers could also be separated at the stage of the formation of the hydroxycholesterol derivatives. We found that the 24-epimers of the Δ^5 -compound can be resolved more easily than that of the corresponding D_3 form. Thus, the 24-epimers of 24-hydroxycholesterol dibenzoate⁵, 24,25-dihydroxycholesterol 3,24-dibenzoate 25-TMS ether^{9,10}, 1α ,24,25-trihydroxycholesterol 1,3,24-tribenzoate 25-TMS ether¹¹ and 1α ,24-dihydroxycholesterol 3,24dibenzoate⁸ could be separated on a silica gel column, and their absolute configurations at the C-24 position were determined^{9-11,17}. In all of these derivatives, the R isomer is less polar than the S isomer.

Although gas chromatography can be used for the identification of D_3 metabolites¹⁸, HPLC is much more useful because it gives sharper peaks and a higher resolution and also can be used for preparative purposes.

These results demonstrate that HPLC is very useful for the identification and determination of D_3 metabolites and their analogues and especially for metabolic studies using labelled compounds.

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